

# M-6316

M.Sc. (I<sup>st</sup> Semester) Examination, 2020

## CHEMISTRY

(Group Theory, Spectroscopy &  
Diffraction Methods)

*Time Allowed : Three Hours*

*Maximum Marks : 70*

*Minimum Pass Marks : 25*

**Note :** Attempt all the five questions, selecting one question from each unit. All questions carry equal marks.

### Unit - I

- Q. 1.** (a) Discuss the identification of unit cells from systematic absences in diffraction pattern. **10**
- (b) The utilized reflecting plane of LiF crystal has a  $d$  value of  $2.014 \text{ \AA}$ . Calculate the wavelength of second order diffracted line which has a value of  $50.1^\circ$ ? **4**

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**P.T.O.**

**(2)**

**Or**

- (a) Derive the following equation : **10**  
 $n\lambda = 2d \sin \theta$
- (b) Discuss Laue's method of X-ray structural analysis of crystals. **4**

### Unit - II

- Q. 2.** (a) Show that symmetry operations of  $C_{3v}$  point group form a mathematical group. **10**
- (b) Determine number of classes of  $C_{3v}$  point group. **4**

**Or**

- (a) Discuss the different rules arises from the solution of orthogonality theorem. **10**
- (b) Explain symmetry elements present in trans- $\text{CH}_2\text{Cl}_2$ . **4**

### Unit - III

- Q. 3.** (a) Discuss the basic theory and instrumentation of fluorescence spectroscopy. **10**
- (b) Write short note on photo-ionization process. **4**

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**(3)**

**Or**

- (a) Explain the various modes of re-emission of energy of a molecule in electronically excited states. **10**
- (b) How will you obtain a photoelectron spectrum ? **4**

**Unit - IV**

- Q. 4.** (a) Write in brief the basic principle of nuclear magnetic resonance spectroscopy. How do NMR and ESR differ in technique and applications. **10**
- (b)  $C^{13}$  is NMR active while  $C^{12}$  is not. Explain. **4**

**Or**

- (a) What do you mean by Coupling Constant ? Discuss the factors affecting coupling constant in detail. **10**
- (b) How many number of signals obtained in the following compounds : **4**

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**P.T.O.**

**(4)**

- (i)  $CH_3COCH_3$
- (ii)  $\begin{array}{c} H_2C - CH_2 \\ | \quad | \\ H_2C - CH_2 \end{array}$
- (iii)  $CH_3OH$
- (iv)  $\begin{array}{c} CH_3 - CH - CH_3 \\ | \\ OH \end{array}$

**Unit - V**

- Q. 5.** (a) Discuss the following : **10**
- (i) Factors affecting the "g" value
- (ii) Zero field splitting
- (b) Explain the rules for predicting number of hyperfine line. **4**

**Or**

Explain the instrumentation, experimental technique and applications of ESR spectroscopy in detail. **14**

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**100**